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STUDY. PLAY. Solute.

Substance that is
dissolved in a solution

Ex. Oxygen in air.

Solvent. The substance
in which the solute
dissolves (substance

that "does" the

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dissolving) Ex. Nitrogen in air. Dissociation. The process in which an ionic compound separates into ions as it dissolves.

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the essential lessons associated with acids, bases and solutions.

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Solutions, Acids

substance that is dissolved in a solvent

to make a solution.

solvent. The substance

in which a solute is

dissolved to form a

solution.

Chapter

8--Solutions, Acids,

& Bases, Chapter 8:

Solutions ...

Chapter 8 Solutions,

Acids, and Bases

Section 8.4 Strength of

Acids and Bases (pages

246-249) This section

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Solutions, Acids

explains how to describe acids and bases in terms of both concentration and strength. Reading Strategy (page 246) Comparing and Contrasting As you read, complete the diagram by comparing and contrasting acids and bases. For more information on

Chapter 8 Solutions, Acids, and Bases Section 8.4 Strength

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Solutions Acids

Ch 8 Solutions, Acids and Bases Study Guide Answers 16. For a solution to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must ATTRACT ONE ANOTHER. 17. During the formation of a solution, energy is ABSORBED & RELEASED. 18. The solubility of solids

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Solutions Acids

increases as the temperature of the solvent INCREASES.

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16. For a solution to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must _____. 17. During the formation of a solution,

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Solutions Acids

energy is _____. 18.

And Bases

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Chapter 8 Solutions,
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Section 8.3 Properties
of Acids and Bases
(pages 240–245) This
section describes the
general properties of
acids and bases.

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Section 8.3 ...

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Chemistry. 65%
average accuracy. 6
months ago, kallen. 0.

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Save. Edit. Edit. ... If 20 cm³ samples of 0.1 mol dm⁻³ solutions of the acids below are taken, which acid would require a different volume of 0.1 mol dm⁻³ sodium hydroxide for complete neutralization?

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Solutions Acids

matthew j. • 19 cards.
solute. a substance
whose particles are
dissolved in a solution.

solvent. the substance
in which the solute
dissolves. dissociation.
the process in which an
ionic compound
separates into ions as
it dissolves. ...

chapter 8 solutions, acids, and bases - Physical Science ...

A chemical reaction
between and acid and

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Solutions Acids

Antibiotics

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a base: Salt: An ionic compound formed when an acid reacts with a base: pH: A measure of the hydronium ion concentration of a solution: Buffer: A solution that is resistant to large changes in pH: Electrolyte: A compound that produces ions when it dissolves

Quia - Chapter 8:

Page 17/25

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Chapter 8

Solutions, Acids, and Bases

Chapter 8, Solutions, Acids, and Bases.

Particles that are dissolved in a solution. The solution in which a solute will dissolve in. The process in which an ionic compound separates into ions as it dissolves. Breaking down into small pieces that spread out through the water.

Quia - Chapter 8,

Page 18/25

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Chapter 8

Solutions, Acids, and Bases

Chapter 8 vocabulary
(Solutions, Acids, and Bases)
8.1- Formation of Solutions Ionization

The process in which neutral molecules gain or lose electrons.

Solute A substance whose particles are dissolved in a solution.

Kool-Aid Dispersion The breaking into the small pieces that

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Chapter 8

Solutions Acids

vocabulary

(Solutions, Acids, and Bases) by ...

solution. 8.3 Properties
of Acids and Bases

Some general
properties of acids
include sour taste,
reactivity with metals,
and ability to produce
color changes in
indicators. • An acid is
a compound that
produces hydronium
ions (H_3O^+) when
dissolved in water. •

An indicator is any

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Solutions, Acids

substance that
changes color in the
presence of an acid or
base.

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Chapter 8 Solutions,
Acids, and Bases

Physical Science

Reading and Study

Workbook Chapter 8 89

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Section 8.1 Formation

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Chapter 8

Solutions Acids

of Solutions (pages 228-234) This section explains the parts of a solution, the processes that occur when

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MSBSHSE Class 8

Science Chapter 12:

Introduction to Acid
and Base Important

Textbook Questions

and Solutions You will

notice that some

substances have a

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sweet taste, some are bitter, some are sour or astringent. Lemon, tamarind, vinegar or amla like substances acquire sour taste due to the presence of a typical compound in them.

MSBSHSE Class 8 Science Chapter 12: Introduction to Acid

...

a chemical reaction between an acid and a base: salt; an ionic

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compound formed
when an acid reacts
with a base: pH: a
measure of the
hydronium ion
concentration of a
solution: buffer: a
solution that is
resistant to large
changes in pH:
electrolyte: a
compound that
produces ions when it
dissolves

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