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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass. Launch Gizmo

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Balancing Chemical Equations. Balance and classify five types of chemical reactions: synthesis, decomposition, single

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replacement, double replacement, and combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and numerical data.

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Balancing Chemical Equations Gizmo Answers Activity B

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This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: $H_2 + O_2 \Rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 2. How many atoms are in an oxygen molecule (O_2)? 2 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? H: 2 O: 1 4. In general, what does a subscript (such as the "2" in H

Balancing Equations Gizmo - Student Exploration Balancing ...

equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the...

Student Exploration- Balancing Chemical Equations (ANSWER ...

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This is the fourth lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions.

Tenth grade Lesson Gizmo: Modeling Chemical Reactions

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