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Cu^{2+} , Fe^{2+} , Fe^{3+} , Mn^{2+} , Co^{2+} , Ni^{2+} Blue Pre. of Cu^{2+} Brown Pre. of Fe^{3+} Green Pre. of Ni^{2+} Pale pink Pre. of Mn^{2+} Pink Pre. of Co^{2+} 2) Noted the solubility a) In water b) In dil.HCl a) Soluble and no ppt with Na_2CO_3 b ...

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Class Xii Chemistry Practical Salt ANALYSIS OF SIMPLE SALT-ANIONS EXPERIMENT ... A little of salt is rubbed with dil H_2SO_4 Conformatory test for Acetate 1) A little of the salt solution is treated with neutral ferric chloride solution 2) A little of salt solution is

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To determine the strength of KMnO_4 solution by titrating it against a standard solution of: 1.Oxalic acid 2.Ferrous ammonium sulphate (Mohr's salt) Preparation of Organic Compounds To prepare; 1.Acetanilide 2.Dibenzal acetone 3.p-Nitroacetanilide 4.2-Naphtholaniline dye

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This is done by conducting a series of tests in a systematic manner and using the observations to confirm the absence or presence of specific cations and anions. Salt analysis is an integral part of the CBSE class 12 chemistry practical examinations and is a topic that several students struggle with.

Salt Analysis: Step-by-Step Guide with Tricks & Shortcuts

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