

## Modern Chemistry Chapter 14 Answers

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### Modern Chemistry Chapter 14 Answers

Modern Chemistry 121 Acids and Bases CHAPTER 14 REVIEW Acids and Bases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Answer the following questions according to the Brønsted-Lowry definitions of acids and bases: \_\_\_\_\_ a.

### Modern Chemistry Chapter 14 Review Answers Acids And Bases

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Modern Chemistry Chapter 14 & 15. binary acid. oxyacid. strong acid. weak acid. an acid that does not contain oxygen, such as hydrofluoric acid. an acid that is a compound of hydrogen, oxygen, and a third ele.... an acid that ionizes completely in a solvent.

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CHAPTER 14 REVIEW Acids and Bases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. a. Write the two equations that show the two-stage ionization of sulfurous acid in water. stage 1:  $\text{H}_2\text{SO}_3(\text{aq}) \rightleftharpoons \text{H}^+(\text{aq}) + \text{HSO}_3^-(\text{aq})$  stage 2:  $\text{HSO}_3^-(\text{aq}) \rightleftharpoons \text{H}^+(\text{aq}) + \text{SO}_3^{2-}(\text{aq})$  b.

### 14 Acids and Bases

Modern Chemistry Chapter 14 Acids & Bases Section 1 Properties of Acids & Bases Aqueous solutions of acids taste sour. Acids change the color of acid-base indicators. Some acids react with active metals and release hydrogen gas. Acids react with bases to produce salts and water. Acids conduct electricity (electrolytes).

### Modern Chemistry Chapter 14

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CHAPTER 14 REVIEW . Acids and Bases. SHORT ANSWER Answer the following questions in the space provided. 1. a. Write the two equations that show the two-stage ionization of sulfurous acid in water. stage 1:  $\text{H}_2\text{SO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{HSO}_3^-(\text{aq})$  b. Which stage of ionization usually produces more ions? Explain your answer.

### CHAPTER 14 REVIEW Acids and Bases

Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company ISBN 978-0-07802-152-7

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Modern Chemistry Chapter 14. boiling-point elevation. colligative properties. dissociation. freezing-point depression. the difference between the boiling point of a pure... solvent and... properties that depend on the concentration of solute... particle.... the separation of ions that occurs when an ionic compound... diss....

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CHAPTER 14 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 14-1 SHORT ANSWER Answer the following questions in the space provided. 1. Use the guidelines in Table 14-1 on page 427 of the text to predict the solubility of the following compounds in water: a. magnesium nitrate b. barium sulfate c. calcium carbonate d. ammonium phosphate

### 14 Ions in Aqueous Solutions and Colligative Properties

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CHAPTER 14 REVIEW Acids and Bases SECTION 2 SHORT ANSWER Answer the following questions in the space provided 1 a Explain your answers BCI 3 is the Lewis acid because it accepts an electron pair in forming a covalent bond It is not a Brønsted-Lowry acid, because it is not donating a proton 122 ACIDS AND BASES MODERN CHEMISTRY Study ...

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CHAPTER 16 REVIEW Reaction Energy SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. For elements in their standard state, the value of  $\Delta H_f^\circ$  is . 2. The formation and decomposition of water can be represented by the following thermochemical equations:  $\text{H}_2(\text{g}) + \frac{1}{2}\text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{g})$  241.8 kJ/mol  $\text{H}_2\text{O}(\text{l})$  241.8 kJ/mol ...

### 16 Reaction Energy

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